



## Nitrous acid in the polluted coastal atmosphere of the South China Sea: Ship emissions, budgets, and impacts



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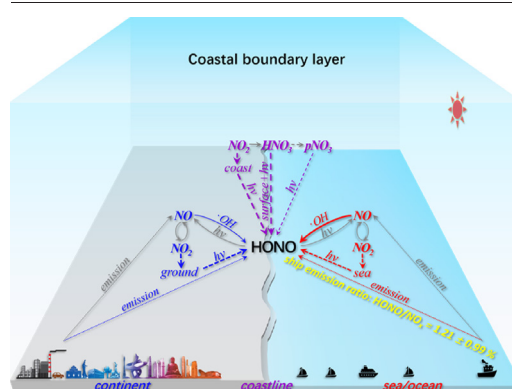
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### HIGHLIGHTS

- The ship emission ratio of HONO/NO<sub>x</sub> is  $1.21 \pm 0.99\%$ .
- The offshore sea is badly impacted by human activities and serves as an HONO sink.
- NO<sub>2</sub> conversion and TNO<sub>3</sub> photolysis dominate HONO production in coastal air.
- NO<sub>x</sub>-related sources dominate HONO production in continental air.
- HONO contributes to increasing AOC in marine, coastal and continental air.

### GRAPHICAL ABSTRACT



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### ABSTRACT

Nitrous acid (HONO) can significantly contribute to hydroxyl radicals ( $\bullet\text{OH}$ ) and thus regulate atmospheric oxidation chemistry; however, ambient HONO sources are not well quantified and vary in different environments. In this study, we conducted comprehensive field observations at a coastal site in the South China Sea and performed chemical box modelling to demonstrate contrasting budgets and impacts of diurnal atmospheric HONO derived from the sea, coastline and continent. The ship emission ratio of HONO/nitrogen oxides (NO<sub>x</sub>) ( $1.21 \pm 0.99\%$ ) was calculated from hundreds of night-time fresh plume measurements. Offshore marine air was frequently influenced by ship exhausts, and the sea acted as an HONO sink. Heterogeneous conversions of nitrogen dioxide (NO<sub>2</sub>) on underlying surfaces and photolysis of adsorbed nitric acid (HNO<sub>3(ads)</sub>) were the major HONO sources in coastal air, when heterogeneous NO<sub>2</sub> conversions on the ground surface and the homogeneous NO +  $\bullet\text{OH}$  reaction dominated HONO formation in continental air. HONO photolysis was a significant source of reactive radicals (RO<sub>x</sub> =  $\bullet\text{OH}$  + HO<sub>2</sub> $\bullet$  + RO<sub>2</sub> $\bullet$ ) in these air masses. Atmospheric box model including only homogeneous HONO source of the NO +  $\bullet\text{OH}$  reactions significantly underpredicted the  $\bullet\text{OH}$  concentration and atmospheric oxidising capacity in coastal and continental air. This study provides new insights into the complex sources and significant impacts of HONO in the polluted coastal boundary layer.

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## 1. Introduction

Nitrous acid (HONO) is a significant contributor to the hydroxyl radicals ( $\cdot\text{OH}$ ), which is the most powerful oxidant initiating daytime atmospheric oxidation processes and affecting regional air quality (Gligorovski et al., 2015; Kim et al., 2014; Stutz et al., 2013; Xue et al., 2016). Despite such profound environmental consequences, ambient HONO sources are not well quantified (Wang et al., 2021; Zhang et al., 2021). Known sources of HONO include the homogeneous reaction of nitric oxide (NO) with  $\cdot\text{OH}$ , heterogeneous conversions of nitrogen dioxide ( $\text{NO}_2$ ) on various surfaces (Ammann et al., 1998; Liu et al., 2020; Liu et al., 2014; Ma et al., 2017; Stemmler et al., 2006; VandenBoer et al., 2013; Yang et al., 2021b), photolysis of particulate nitrate ( $\text{pNO}_3$ ) and surface-adsorbed nitric acid ( $\text{HNO}_3$  (ads)) (Romer et al., 2018; Ye et al., 2016a; Ye et al., 2017; Ye et al., 2016b; Zhou et al., 2011), and primary emissions from fuel combustion and soil-air exchange (Liang et al., 2017; Liao et al., 2021; Oswald et al., 2013; Peng et al., 2020; Wang et al., 2021; Wormhoudt et al., 2007; Xue et al., 2021). However, some key parameters and processes of these known HONO generation pathways are highly variable and uncertain (Liu et al., 2019; Xue et al., 2020). In particular, HONO emissions from ship exhausts have received limited attention, despite the rapid development of the global shipping industry and the dense maritime traffic in many coastal areas (United Nations Conference On Trade and Development, 2020; Liu et al., 2016). Recently, the  $\Delta\text{HONO}/\Delta\text{NO}_x$  (where  $\text{NO}_x$  = nitrogen oxides) emission ratio of a single medium-speed diesel engine vessel during its cruise was calculated to be  $0.51 \pm 0.18\%$ , and a global ship HONO emission inventory was developed (Sun et al., 2020). Model simulations show that ship-contributed HONO could enhance the concentrations of fine particulate matter ( $\text{PM}_{2.5}$ ), ozone ( $\text{O}_3$ ) and reactive radicals ( $\text{RO}_x = \cdot\text{OH} + \text{HO}_2\cdot + \text{RO}_2\cdot$ , where  $\text{HO}_2\cdot/\text{RO}_2\cdot$  are the hydro/organic peroxy radicals) over the South China Sea by 1%, 6% and 12%, respectively (Dai and Wang, 2021).

Ocean covers approximately 71% of Earth's surface, and over 60% of the global population is concentrated in coastal areas. In the coastal boundary layer, atmospheric compositions and chemical processes differ depending on whether air originates from the sea, coastline or continent (Crowley et al., 2011). Most HONO researches have been dedicated to understanding its sources and impacts in human-perturbed inland areas (Fu et al., 2019; Liu et al., 2021; Liu et al., 2019), whereas few studies have investigated HONO chemistry in marine and coastal atmospheres.

The nocturnal conversion rate of  $\text{NO}_2$  to HONO in marine air was found to be significantly larger than that in air from the land, suggesting oceanic HONO production (Yang et al., 2021a; Zha et al., 2014). Wen et al. (2019)

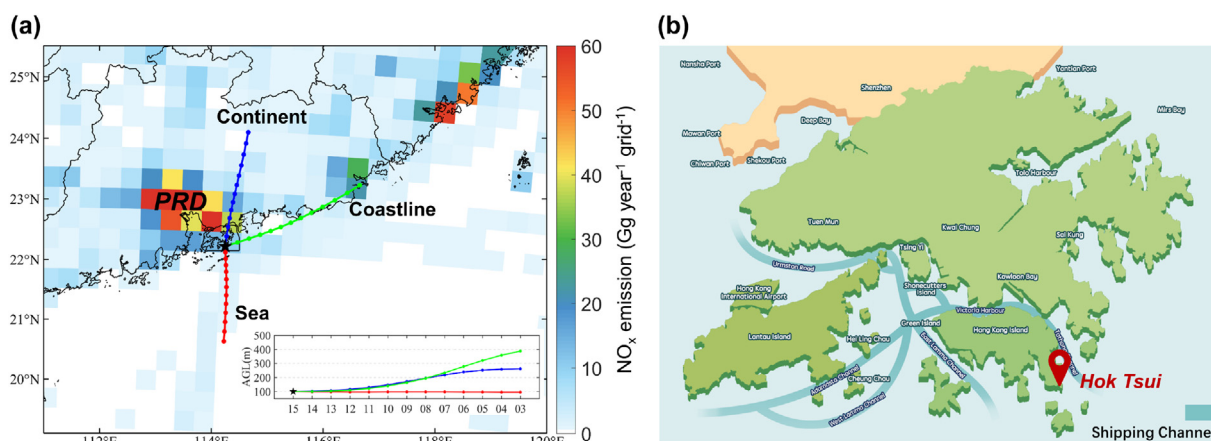
proposed that HONO production in the marine boundary layer might result from air-sea interaction and photochemical formation in the sea-surface microlayer, contradicting the previously accepted hypothesis that the sea surface serves as an HONO sink due to its alkaline nature (Beine et al., 2005; Wojtal et al., 2011). A recent airborne study found rapid reactive nitrogen recycling in the marine atmosphere of the North Atlantic Ocean and suggested that  $\text{pNO}_3$  photolysis is a significant HONO source in this region (Ye et al., 2016b), in contrast to the  $\text{NO}_x$ -dominated HONO formation process in polluted continental areas. Overall, there is incomplete knowledge of marine HONO sources and processes, meaning that investigations are needed on HONO sources in marine air, anthropogenic activities and marine atmospheric processes interactions, and their impacts on coastal air quality.

Hong Kong lies adjacent to the South China Sea and the Northwest Pacific Ocean and is located in the southeast of the Guangdong-Hongkong-Macao Greater Bay Area, one of the world's largest urban agglomerations (Fig. 1). Driven by the Asian monsoon and sea-land breezes, Hong Kong receives air masses from multiple sources, including relatively clean marine air, polluted air masses derived from ship exhausts from surrounding navigation waters, and transported continental air (Wang et al., 2009). Therefore, Hong Kong is a suitable location to study HONO chemistry in dissimilar air masses derived from different environments. We therefore conducted intensive field experiments at a site on the southeast coast of Hong Kong in 2018, measuring a suite of chemical and meteorological parameters. These data facilitated a comprehensive HONO source analysis in four distinct air masses (ship exhausts, marine, coastal and continental air masses) and detailed modelling of the impacts of HONO chemistry on atmospheric oxidising capacity (AOC). Our findings enrich understanding of the complex sources and significant impacts of HONO in the coastal boundary layer and support the development of a refined inventory of HONO emissions from ships.

## 2. Methodology

### 2.1. Field measurements

We conducted field measurements at the Cape D'Aguilar Supersite Air Quality Monitoring Station ( $22^\circ 13' \text{N}$ ,  $114^\circ 15' \text{E}$ ), which is managed by the Hong Kong Environmental Protection Department. This typical coastal background station sits on the southeastern tip of Hong Kong island and is 60 m above sea level (see Fig. 1), a site that receives a variety of air masses, including marine air, continental air and ship plumes (Wang et al., 1998; Wang et al., 2009). Two intensive observation campaigns



**Fig. 1.** Maps showing the study area and observation site. (a) Position of the measurement site in the Guangdong-Hongkong-Macao Greater Bay Area, and the 12-h backward trajectory clusters of air masses coming from three typical coastal environments of the sea, coastline and continent. The map is color-coded with anthropogenic  $\text{NO}_x$  emission intensity. (b) Location of the measurement site of Hok Tsui and the major shipping channels in Hong Kong. Anthropogenic  $\text{NO}_x$  emission data in grids of  $27 \text{ km} \times 27 \text{ km}$  are obtained from the MEIC (2017) (<http://meicmodel.org/>, last access 26 February) for mainland China (Liu and Wang, 2020), the MIX (2010) for Hong Kong, and the HTAP (2010) for the ocean area (Janssens-Maenhout et al., 2015).

were conducted from 31 August to 8 October 2018 and from 31 October to 12 December 2018. Online measurements included HONO and related trace gases (e.g., NO, NO<sub>2</sub>, total reactive oxidised nitrogen (NO<sub>y</sub>), O<sub>3</sub>, carbon monoxide (CO), and sulfur dioxide (SO<sub>2</sub>)), 1-, 2.5- and 10- $\mu$ m particulate matters (PM<sub>1</sub>, PM<sub>2.5</sub> and PM<sub>10</sub>), water-soluble inorganic ions in PM<sub>2.5</sub> and PM<sub>10</sub>, volatile organic compounds (VOCs), aerosol surface area density (S<sub>a</sub>) and meteorological parameters (e.g., solar radiation, temperature, relative humidity (RH), wind direction and wind speed).

HONO was measured using a custom-built iodide-anion chemical-ionization mass spectrometer (CIMS; THS Instruments Inc., Atlanta) that was optimised to measure HONO (Wang et al., 2016; Wang, 2020). Briefly, HONO was ionised and combined with iodide anion to form iodide-HONO clusters (IHONO<sup>-</sup>) and detected by a quadrupole mass spectrometer. During one measurement period (31 August–4 September 2018), we simultaneously measured HONO using the CIMS and a long-path absorption photometer (LOPAP-03; QUMA, Germany), a widely used commercial instrument. Two sets of measured HONO data agreed well, with a correlation coefficient ( $r^2$ ) of 0.75 (Fig. S1). Detailed description of the HONO measurement using CIMS is provided in Text S1 in the Supplementary Information (SI). The other species and parameters were measured using commercial instruments; detailed measurement information is documented in Table S1.

## 2.2. Chemical box model simulation

We used a chemical box model equipped with the Master Chemical Mechanisms (MCM, version 3.3.1) to quantify the roles of HONO in atmospheric radical productions and AOC. The MCM box model comprises almost explicit gaseous chemical mechanisms and has been widely used in previous researches (Jiang et al., 2020; Xue et al., 2016). The model was constrained by observational data (concentrations of CO, SO<sub>2</sub>, O<sub>3</sub>, NO, NO<sub>2</sub>, VOCs, and meteorological parameters) averaged at 10-min time intervals for all but VOCs, which were collected at 1-h intervals. Before each formal simulation, we conducted a three-day spin-up constrained by observational data to reach stable states for the unconstrained species. In each case, two runs with/without measured HONO constraint were conducted to reveal the impacts of HONO sources besides the NO + •OH reaction on atmospheric oxidation chemistry. The model configuration details are provided in Text S2.

## 3. Results and discussion

### 3.1. Overview of observation data

Timeseries of the atmospheric trace compositions and meteorological conditions measured at the site are presented in Fig. S2 (10-min data-point intervals, except 1-h data-point intervals for HNO<sub>3</sub> and pNO<sub>3</sub>). Overall, the observation results revealed highly variable moderate air pollution. The 10-min mean  $\pm$  standard deviation (SD) concentration of HONO was 0.18  $\pm$  0.15 ppbv. This is significantly lower than those at suburban coastal sites in eastern and southern China and a strait in Canada (Cui et al., 2019; Wojtal et al., 2011; Xu et al., 2015; Yang et al., 2021a), comparable to that in the inland Bohai Sea in northern China (Wen et al., 2019), 50% higher than that measured 6 years ago at a site near the present station (Zha et al., 2014), and substantially higher than those in more remote marine boundary layers in the North Atlantic Ocean and the Mediterranean Sea (Meusel et al., 2016; Ye et al., 2016b). This intermediate concentration of HONO is consistent with the moderate concentrations of its precursors, such as NO (0.66  $\pm$  1.83 ppbv), NO<sub>2</sub> (4.28  $\pm$  4.38 ppbv), HNO<sub>3</sub> (0.25  $\pm$  0.12 ppbv) and pNO<sub>3</sub> (3.3  $\pm$  1.7  $\mu$ g m<sup>-3</sup>, concentration in PM<sub>10</sub>).

### 3.2. Ship emission of HONO

#### 3.2.1. Ship plumes selection

The timeseries (Fig. S2) show frequent short-term concentration spikes of SO<sub>2</sub> and NO<sub>x</sub>, indicating impacts from ship plumes from the dense

maritime traffic in Hong Kong's waters and the surrounding waters (Wang et al., 2003). To examine the emission ratios of HONO in ship exhausts, we adopted a set of criteria to screen out ship plumes (Table S2). First, only data within the highest 25% of NO<sub>x</sub> and SO<sub>2</sub> concentrations were considered. Second, data with short-term (<2 h) SO<sub>2</sub> concentration spikes were identified as ship plumes (SPs), and those with NO<sub>x</sub>/NO<sub>y</sub> ratios  $\geq$  90% were considered fresh ship plumes (FSPs). Eventually, we selected data from 567 SPs and 272 FSPs based on these criteria. Fig. 2a shows representative examples of identified SPs, and Fig. S2 shows all of the SPs observed in this study. Data within the lowest 5% of NO<sub>x</sub> and PM<sub>2.5</sub> concentrations were considered background air characteristics, and 13 background cases (BGCs) were identified and used to remove the influence of background concentrations from calculations of ship emission ratios (see next section). Table S3 and Fig. 4 show statistics of the measured species and parameters in the ship plumes and BGCs. HONO concentrations were lowest in the BGCs (0.01  $\pm$  0.01 ppbv), higher in FSPs (0.17  $\pm$  0.21 ppbv) and highest in SPs (0.23  $\pm$  0.22 ppbv), with corresponding HONO/NO<sub>x</sub> ratios of 0.012  $\pm$  0.016, 0.011  $\pm$  0.009 and 0.022  $\pm$  0.021, respectively. These notably enhanced HONO concentrations in FSPs and SPs compared to BGCs indicate that ships release high concentrations of HONO. The higher HONO concentrations and HONO/NO<sub>x</sub> ratios in SPs than in FSPs suggest there is considerable HONO production during the ageing of FSPs.

#### 3.2.2. Ship emission ratio of HONO/NO<sub>x</sub>

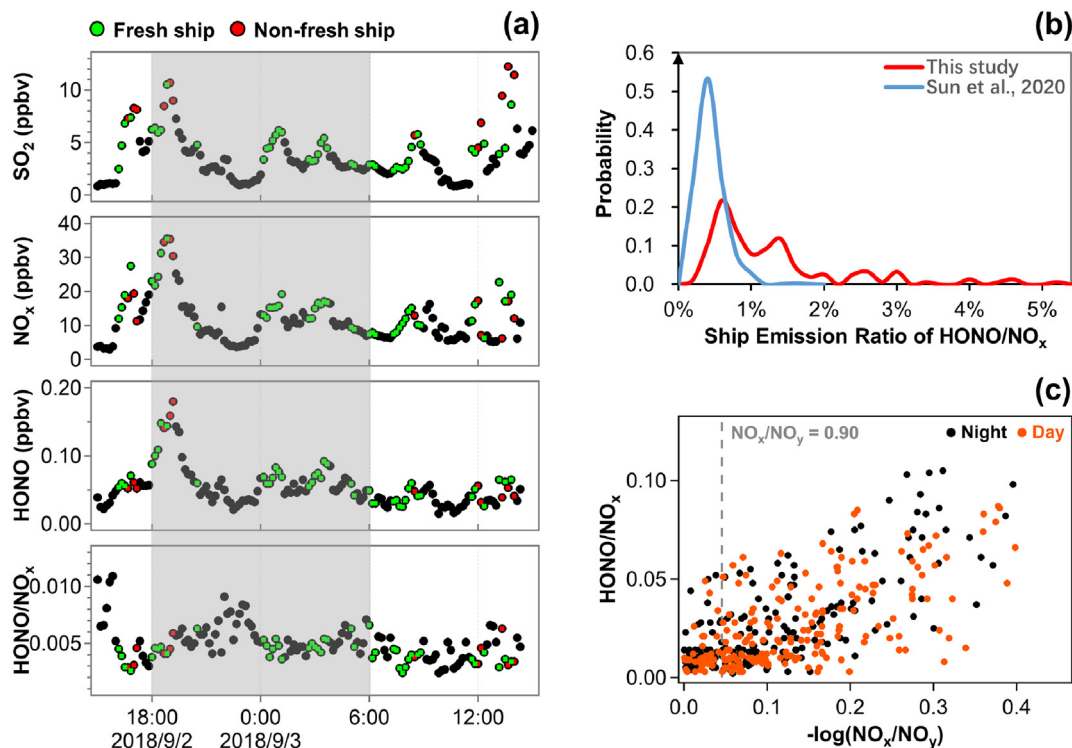
The HONO/NO<sub>x</sub> emission ratio (ER<sub>HONO/NO<sub>x</sub></sub>) is a key parameter for evaluating the characteristics and intensity of HONO emission from ships. ER<sub>HONO/NO<sub>x</sub></sub> was deduced in this study using Eq. (1).

$$\text{HONO/NO}_x = \frac{\text{HONO}_{(\text{FSP})} - \text{HONO}_{(\text{BGC})}}{\text{NO}_{x(\text{FSP})} - \text{NO}_{x(\text{BGC})}} \quad (1)$$

where HONO<sub>(FSP)</sub> and NO<sub>x(FSP)</sub> are the HONO and NO<sub>x</sub> concentrations in FSPs, and HONO<sub>(BGC)</sub> and NO<sub>x(BGC)</sub> are the average HONO and NO<sub>x</sub> concentrations in BGCs, respectively. To negate the influence of light on HONO chemistry, only night-time FSPs ( $n = 151$ ) were included in the ER<sub>HONO/NO<sub>x</sub></sub> calculation. The ship ER<sub>HONO/NO<sub>x</sub></sub> varied from 0.21% to 5.30% (75% within 0.20–1.40% and 25% > 1.40%) with a mean  $\pm$  SD of 1.21  $\pm$  0.99% and a median of 0.91% (Fig. 2b). This large variation in ship ER<sub>HONO/NO<sub>x</sub></sub> is similar to that of vehicles and is due to the diversity of engine types, fuels, operation conditions and other factors (Kurtenbach et al., 2001; Liang et al., 2017; Nakashima and Kajii, 2017). As shown in Fig. 2b, the ER<sub>HONO/NO<sub>x</sub></sub> obtained in this study is more than double and more variable than that reported by Sun et al. (2020) (mean  $\pm$  SD, 0.51  $\pm$  0.18%; median, 0.43%; 90% within 0.08–0.97%). The emission ratios in Sun et al. (2020) were derived from only one small vessel (with a gross tonnage of 3235 tons) equipped with a medium-speed engine and fuelled by high-quality light marine diesel oil. In contrast, the present study was conducted near a navigation area in one of the world's busiest harbour cities. The diversity of vessels and complexity of ambient conditions in the study area accounts for the much higher and broader ship ER<sub>HONO/NO<sub>x</sub></sub> than that of Sun et al. (2020).

#### 3.2.3. Ship plumes ageing process

The HONO/NO<sub>x</sub> ratio as a function of  $-\log_{10}(\text{NO}_x/\text{NO}_y)$ , an indicator of photochemical age (Kleinman et al., 2008; Nie et al., 2012), is illustrated in a scatter plot in Fig. 2c. As ship plumes age, their HONO/NO<sub>x</sub> ratio gradually increases. For example, as  $-\log_{10}(\text{NO}_x/\text{NO}_y)$  increased from 0.05 to 0.30, the NO<sub>x</sub>/NO<sub>y</sub> ratio decreased from 0.90 to 0.50, and the average HONO/NO<sub>x</sub> proportion rose from 1.64% to 5.79%, tripling from 1.54% to 4.57% in the daytime and quadrupling from 1.78% to 7.15% at night. These rising trends show that there is HONO formation during ship plume ageing, confirming that ship emissions are a significant source of HONO in coastal and maritime areas with abundant shipping traffic.



**Fig. 2.** Ship emission of HONO. (a) Examples of ship plumes in the time series of SO<sub>2</sub>, NO<sub>x</sub>, HONO and HONO/NO<sub>x</sub>. Grey shaded area marks the night period. (b) Probability histograms of the ship emission ratio of HONO/NO<sub>x</sub> in this study and Sun et al. (2020). (c) The ratio of HONO/NO<sub>x</sub> against the ageing process of all ship plumes during the day and night.

### 3.3. Contrasting characteristics, chemical budgets and impacts of HONO in marine, coastal and continental air masses

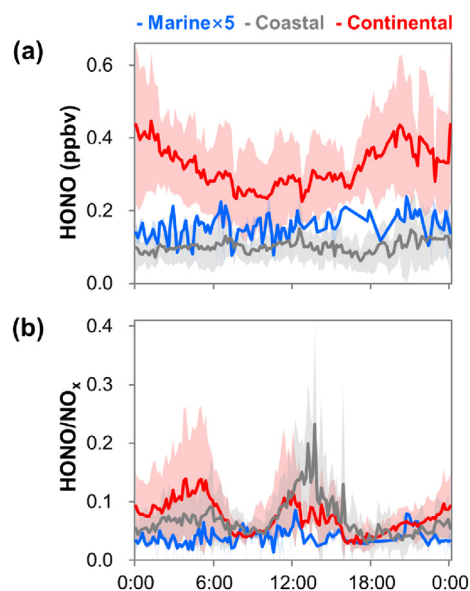
#### 3.3.1. Selection of different air masses

Three types of air masses – originating from the sea, coastline and continent – were selected by examining backward trajectory and wind information. Within the 24 h of a day, when all 12-h backward trajectories (computed hourly using the HYSPLIT model) and the average wind were both from the sea (or coastline, continent) (Stein et al., 2015), the day was regarded as a marine (or coastal, continental) air-dominated case. With such strict criteria, 4, 11 and 7 days were classified as marine, coastal and continental air cases, respectively (see Fig. 1 for the three types of trajectory clusters and Fig. S2 for the individual cases). Air masses from the sea were frequently influenced by ship plumes, as indicated by their spikes in SO<sub>2</sub> and NO<sub>x</sub> concentrations. For this reason, we applied more rigorous screening criteria to the marine air cases: only marine air data within the lowest 50% of SO<sub>2</sub> and NO<sub>x</sub> concentrations were selected for further analysis (194 data points).

#### 3.3.2. Air mass characteristics

Fig. 4 shows the average physical and chemical features of the three types of air masses. Marine air was very humid (RH = 93 ± 3%), with the highest freshness (NO<sub>x</sub>/NO<sub>y</sub> = 0.89 ± 0.19) and lowest air pollutant concentrations. Coastal air was humid (RH = 82 ± 4%), with the highest degree of ageing (NO<sub>x</sub>/NO<sub>y</sub> = 0.53 ± 0.23) and moderate air pollutant concentrations, except for that of pNO<sub>3</sub>, which was the highest in coastal air. Continental air was relatively dry (RH = 63 ± 14%), with a moderate degree of ageing (NO<sub>x</sub>/NO<sub>y</sub> = 0.75 ± 0.21) and the highest concentrations of air pollutants, especially NO<sub>x</sub>. The average HONO concentrations were 0.03 ± 0.01, 0.10 ± 0.05 and 0.32 ± 0.14 ppbv in marine, coastal and continental air, respectively, with respective HONO/NO<sub>x</sub> ratios of 0.008 ± 0.003, 0.067 ± 0.054 and 0.074 ± 0.065.

The diurnal variations in HONO and the HONO/NO<sub>x</sub> ratio are shown in Fig. 3. The HONO concentration presented a typical diel pattern in continental air, reaching a maximum at approximately midnight and a minimum at noon, whereas concentrations remained relatively stable throughout the day in marine and coastal air. Considering the rapid photolytic loss of



**Fig. 3.** Averaged diurnal variations of (a) HONO and (b) HONO/NO<sub>x</sub> in air masses coming from the sea (Marine), coastline (Coastal) and continent (Continental). Shaded areas show standard deviations (±1σ). Note that the values of HONO and HONO/NO<sub>x</sub> in marine air are magnified five time.

HONO in the daytime, especially at noon, the stable HONO concentration throughout the day suggests that HONO is sufficiently emitted and/or generated in marine and coastal air. The HONO/NO<sub>x</sub> ratio varied slightly in marine air remaining approximately 0.008, but fluctuated sharply showing two peaks before sunrise and around noon in both coastal (0.092 and 0.232) and continental (0.139 and 0.118) air. While the dawn peak was ascribable to night-time HONO accumulation, the noon peak could be due to enhanced photolytic sources. The HONO/NO<sub>x</sub> ratio increased sharply at approximately noon in coastal air, to a level twice that in continental air, suggesting that there was a greater increase in photo-related HONO production in coastal air than in continental air.

### 3.3.3. HONO budgets

We conducted a HONO budget analysis to explore the contributions of different sources in marine, coastal and continental air (Liu et al., 2021; Oswald et al., 2015; Wang et al., 2017; Xue et al., 2020). We considered HONO derived from on-road vehicular emission, the homogeneous NO + •OH reaction, light-sensitive heterogeneous NO<sub>2</sub> conversion on terrestrial, sea and aerosol surfaces and HNO<sub>3(ads)</sub> and pNO<sub>3</sub> photolysis, which have been reported as potentially significant HONO contributors. HONO from other sources (e.g., HO<sub>2</sub>-H<sub>2</sub>O + NO<sub>2</sub> reaction, NO<sub>2</sub> + H<sub>2</sub>O reaction and ortho-nitrophenol photolysis) were not considered because their reaction rate constants were negligible or there were no relevant data available. The HONO sinks included in the analysis were photolysis, the HONO + •OH reaction and dry deposition. More details of the budget analysis method are provided in the SI (Text S3). Note that some parameters (e.g., emission ratios, reaction rate constants and precursor uptake coefficients) used for the budget calculation are subject to uncertainties. We adopted the commonly used values in our analysis and performed sensitivity tests to examine their impacts on the results.

Fig. 5 shows the average daytime (09:00–17:00, local time (LT)) production rates of HONO (P<sub>HONO</sub>) calculated from individual sources [P<sub>HONO(cal)</sub>] and deduced from observations [P<sub>HONO(obs)</sub>], where P<sub>HONO(obs)</sub> equals the sum of the observed changing rate and total calculated loss rate of HONO. The P<sub>HONO(obs)</sub> values in marine, coastal and continental air were 0.06 ± 0.01, 0.25 ± 0.08 and 0.82 ± 0.24 ppbv h<sup>-1</sup>, respectively. Continental air masses pass through the Pearl River Delta urban agglomerations (Fig. 1); thus, their higher

P<sub>HONO(obs)</sub> value reflects the significant effects of human activities on HONO chemistry in the coastal boundary layer of southern China. The budget analysis shows that P<sub>HONO(cal)</sub> was much higher than P<sub>HONO(obs)</sub> in marine air but matched the daytime variations of P<sub>HONO(obs)</sub> in coastal and continental air.

The source contributions to HONO differed between the three air mass types. In marine air (Fig. 5a), the homogeneous NO + •OH reaction (45.8 ± 6.7%) and the heterogeneous NO<sub>2</sub> conversion on the sea surface (39.0 ± 2.9%) dominated daytime HONO formation, both of which far exceeded the P<sub>HONO(obs)</sub>. Note that we have assumed the same NO<sub>2</sub> uptake coefficient and conversion efficiency on the sea surface as on the land surface. The large contributions of NO<sub>x</sub> to in-situ HONO production suggest that ship emissions strongly affected the selected marine-air cases. P<sub>HONO(cal)</sub> was higher than P<sub>HONO(obs)</sub>, implying that compared to other areas, there were larger HONO losses in the marine boundary layer, where more efficient HONO uptake by the alkaline sea may occur.

In coastal air (Fig. 5b), HNO<sub>3(ads)</sub> and pNO<sub>3</sub> photolysis were the most significant sources of HONO (38.5 ± 3.7% totally) around noon (11:00–14:00 LT), and the heterogeneous NO<sub>2</sub> conversion on the underlying surface governed HONO formation throughout the remainder of the day (40.7 ± 5.4% throughout the daytime). Coastal air featured moderate NO<sub>2</sub> concentrations and abundant secondary pollutants, such as total oxidised nitrogen (TNO<sub>3</sub> = HNO<sub>3</sub> + pNO<sub>3</sub>) (Fig. 4). Efficient TNO<sub>3</sub> renoxification to nitrogen oxides, in which HONO is an intermediate that readily photolyses to NO, could greatly promote the recycling of TNO<sub>3</sub>-HONO-NO<sub>x</sub> in coastal air.

In continental air (Fig. 5c), NO<sub>x</sub>-related sources – the heterogeneous conversion of NO<sub>2</sub> to HONO on the ground surface (54.7 ± 12.2%) and the NO + •OH reaction (25.5 ± 9.7%) – were the major contributors to daytime HONO formation. The contribution of the NO + •OH reaction reached a maximum in the morning (09:00–10:00 LT; 39.2 ± 2.0%), and the heterogeneous conversion of NO<sub>2</sub> on the ground surface showed increasing contributions as the day progressed. HNO<sub>3(ads)</sub> photolysis (11.0 ± 3.5%) and on-road vehicle emissions (7.4 ± 4.7%) were also significant sources of daytime HONO; the contributions from on-road vehicular emissions reached a peak in the morning and evening and reached a valley at approximately noon, whereas HNO<sub>3(ads)</sub> photolysis exhibited an opposite daytime trend.

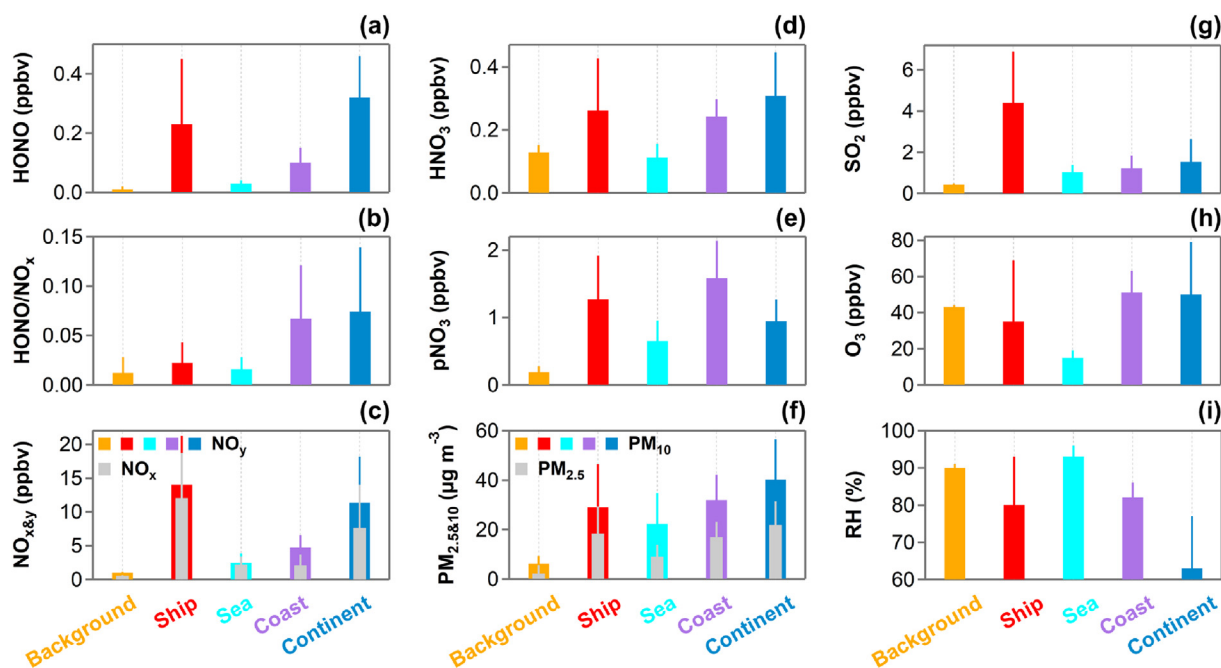


Fig. 4. Atmospheric chemical compositions in different types of plumes. Average concentrations of (a) HONO, (b) HONO/NO<sub>x</sub>, (c) NO<sub>x&y</sub>, (d) HNO<sub>3</sub>, (e) pNO<sub>3</sub>, (f) PM<sub>2.5&10</sub>, (g) SO<sub>2</sub>, (h) O<sub>3</sub>, and (i) RH (%) in the five types of air masses, namely, background, ship, marine, coastal, and continental air. Error bars show the standard deviations (±1σ).

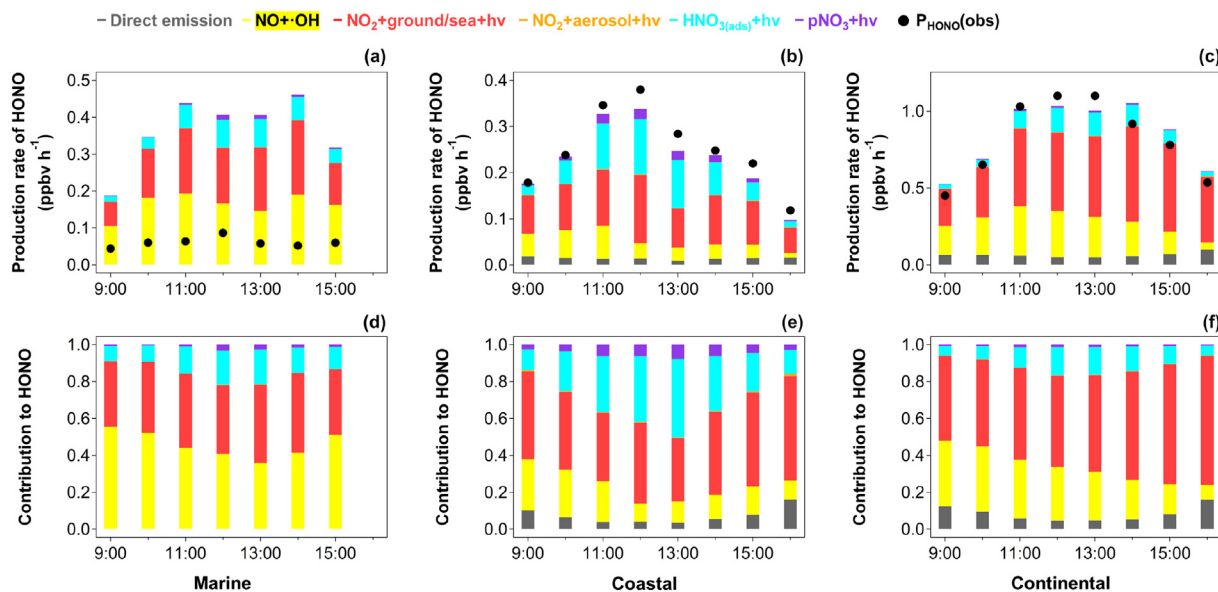


Fig. 5. Average daytime variations in the production rates of HONO calculated from individual sources (bars, upper panel) and deduced from observations and sinks (dots, upper panel) and the contribution of each source (lower panel) in different air masses coming from the sea (left), coastline (middle) and continent (right).  $P_{\text{HONO}}(\text{obs})$  equals the sum of the observed changing rate and the calculated total loss rate, of HONO.

To assess the uncertainties of the source contributions, we performed sensitivity analyses using both upper and lower values for individual parameters (Table S4), which were the vehicular  $\text{HONO}/\text{NO}_x$  emission ratio ( $ER_{\text{HONO}/\text{NO}_x}$ ), the  $\text{NO}_2$  uptake coefficients on various surfaces ( $\gamma_{\text{NO}_2}$ ), and the photolysis frequencies of  $\text{HNO}_{3(\text{ads})}$  [ $j(\text{HNO}_{3(\text{ads})})$ ] and  $\text{pNO}_3$  [ $j(\text{pNO}_3)$ ]. Fig. 6 shows the average diurnal sensitivity results. Overall, the uncertainties caused by vehicular  $ER_{\text{HONO}/\text{NO}_x}$ ,  $\gamma_{\text{NO}_2}$  on aerosol surfaces and  $j(\text{pNO}_3)$  were small, mostly because of their low contributions to HONO production at our site. The major uncertainties were detected in the parameterisations of  $\gamma_{\text{NO}_2}$  on the ground surface and  $j(\text{HNO}_{3(\text{ads})})$ , which were key parameters of the two principal HONO formation pathways here. Adjusting  $\gamma_{\text{NO}_2}$  on the ground surface led to an HONO production rate far beyond the  $P_{\text{HONO}}(\text{obs})$  or lower than the  $P_{\text{HONO}}(\text{NO} + \cdot\text{OH})$  in all air types. Similarly, adjusting  $j(\text{HNO}_{3(\text{ads})})$  produced a HONO production rate far outside the range between  $P_{\text{HONO}}(\text{obs})$  and  $P_{\text{HONO}}(\text{NO} + \cdot\text{OH})$  in marine and coastal air. However, in continental air, the budget result was not so

sensitive to  $j(\text{HNO}_{3(\text{ads})})$  adjustment because of the limited contribution of  $\text{HNO}_{3(\text{ads})}$  photolysis to HONO production; the abundant fresh exhausts (e.g.,  $\text{NO}_x$ ) indicate the significant roles of on-road emission and  $\text{NO}_2$  conversions in HONO production, and the small proportion of secondary pollutants (e.g.,  $\text{HNO}_3$ ) suggests the limited contribution of high-oxidation-state nitrogen photolysis to HONO. The sensitivity analysis results suggest there were significant uncertainties in HONO production from individual sources, which implies that our quantifications of HONO sources were also of uncertainty. Thus, more accurate parameterisations of these sources are needed because these parameters are fundamental to achieving a comprehensive understanding of HONO chemistry in various environments.

### 3.3.4. HONO impacts on AOC

Fig. 7 shows the average daytime variations in the primary production rates of  $\text{RO}_x$  radicals in the three types of air masses. The daytime

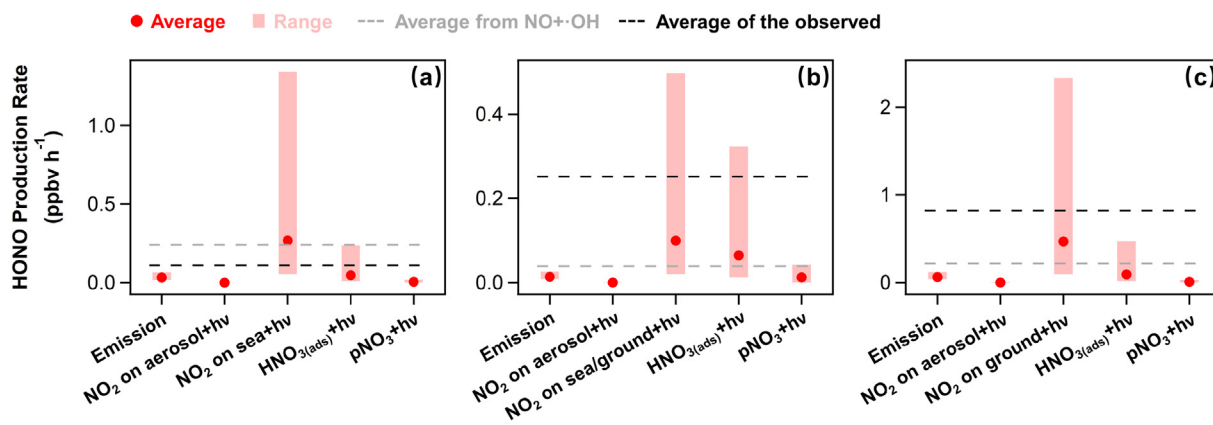
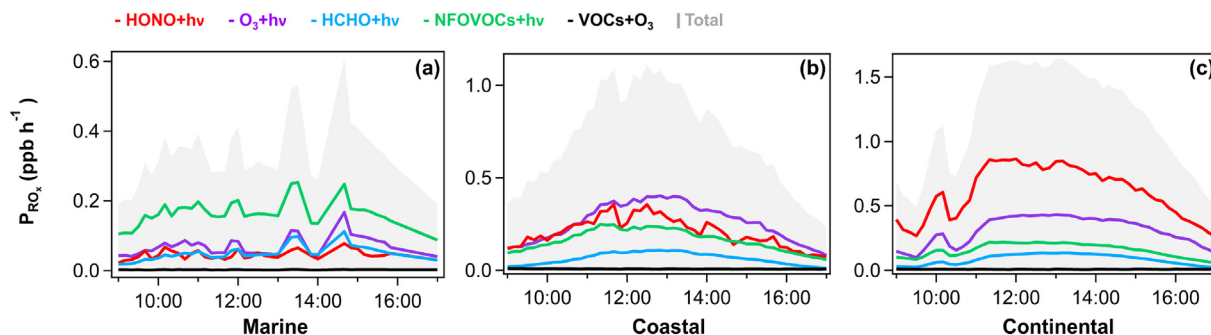


Fig. 6. Sensitivity results of diurnal (09:00–17:00) HONO production rates ( $P_{\text{HONO}}$ ) from different sources in (a) marine, (b) coastal and (c) continental air. The considered HONO sources include direct emission, photo-induced heterogeneous conversion of  $\text{NO}_2$  on aerosol/ground (or sea) surfaces, photolysis of adsorbed  $\text{HNO}_3$  ( $\text{HNO}_{3(\text{ads})}$ ), and photolysis of particulate nitrate ( $\text{pNO}_3$ ). The red dot is the average  $P_{\text{HONO}}$  from each source, the red column indicates the sensitivity range of  $P_{\text{HONO}}$  from each source, the black dashed line is the average observed  $P_{\text{HONO}}$ , and the grey dashed line represents the  $P_{\text{HONO}}$  from  $\text{NO} + \cdot\text{OH}$ .



**Fig. 7.** Average daytime primary production rates of reactive radicals ( $RO_x = \cdot OH + HO_2\cdot + RO_2\cdot$ ) in air masses coming from the (a) sea, (b) coastline and (c) continent in the coastal boundary layer of the PRD. Here the considered major primary radical sources include photolysis of HONO,  $O_3$ , formaldehyde (HCHO), and OVOCs except for HCHO (NFOVOCs), and the reactions of  $O_3$  with unsaturated VOCs.

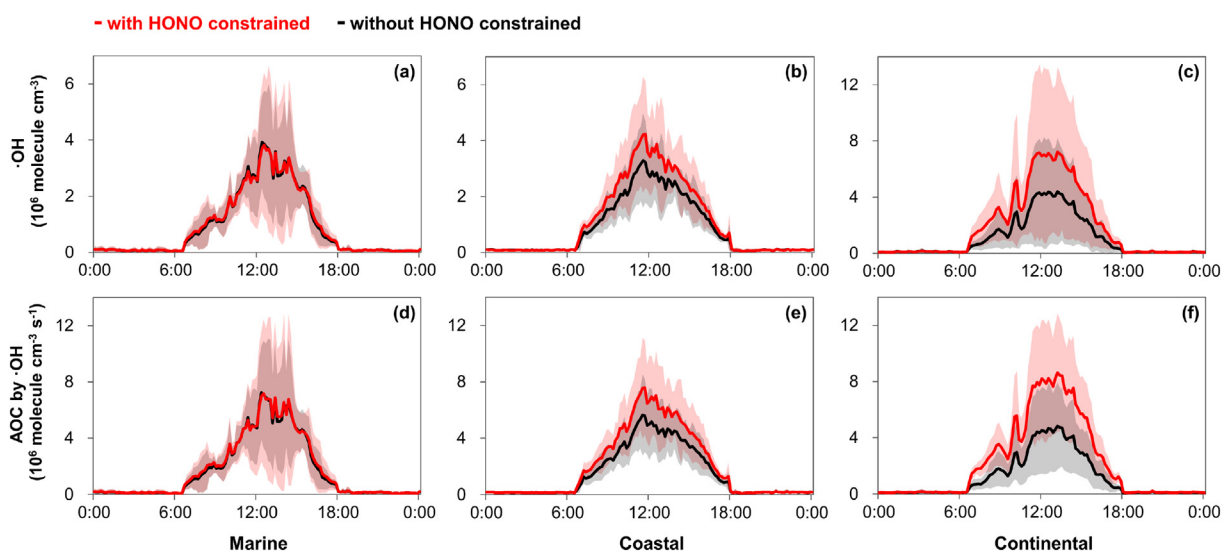
(09:00–17:00 LT)  $RO_x$  production rates were  $0.34 \pm 0.09$ ,  $0.71 \pm 0.26$  and  $1.17 \pm 0.39$   $ppb\ h^{-1}$  in marine, coastal and continental air, respectively. In marine air, the photolysis of oxygenated VOCs (OVOCs, including formaldehyde (HCHO)) was the dominant source of radicals (64.7%; 14.9% for HCHO alone), followed by  $O_3$  photolysis (20.6%), HONO photolysis (14.7%) and the negligible ozonolysis reactions of unsaturated VOCs. In coastal air, the predominant sources of radicals were the photolysis of  $O_3$  (36.6%), OVOCs (32.4%; 9.1% for HCHO alone) and HONO (29.6%), and then the reactions of  $O_3 + VOCs$  (1.4%). In continental air, the photolysis of HONO was the dominant radical source (52.1%), followed by the photolysis of  $O_3$  (26.5%) and OVOCs (20.5%; 7.3% for HCHO alone) and reactions of  $O_3 + VOCs$  (0.9%).  $RO_x$  concentrations increased from marine to coastal to continental air (Fig. S3), and the contribution of HONO photolysis to  $RO_x$  production followed a similar increasing trend. These results suggest that HONO plays a significant role in atmospheric oxidation in the coastal boundary layer, especially in polluted continental environments.

As the basic MCM model (version 3.3.1) used here only includes homogeneous HONO formation from the  $NO + \cdot OH$  reaction, we assessed the magnitudes of  $\cdot OH$  radical and AOC underprediction using the default HONO scheme. AOC was defined as the sum of reaction rates between the  $\cdot OH$  radical and reduced substances (i.e., CO,  $CH_4$ , VOCs, NO,  $NO_2$  and  $SO_2$ ) (Xue et al., 2016). We conducted two runs with/without the constraint of observed HONO data; the difference between the simulation

results represents the contributions of HONO sources except for the  $NO + \cdot OH$  reaction. Fig. 8 shows that the model run with only  $NO + \cdot OH$  as the HONO source compared well with HONO-constrained simulation for marine air but significantly underestimated the  $\cdot OH$  concentrations and AOCs of coastal and continental air. The former can be explained by the underestimation of HONO loss rates in marine air, and the latter by the important role of heterogeneous and photolytic HONO sources in coastal and continental air, as reflected in the budget analysis results. Quantitatively, underestimations by model runs including only  $NO + \cdot OH$  reaction were  $21.4 \pm 2.2\%$  and  $24.8 \pm 2.1\%$  for the average daytime  $\cdot OH$  concentration and AOC in coastal air, respectively, further increasing to  $42.4 \pm 2.4\%$  and  $47.3 \pm 2.8\%$  for continental air. The magnitudes of underestimates are comparable to those reported in other studies performed in mountainous and coastal regions (Jiang et al., 2020; Yang et al., 2021a). These results demonstrate that atmospheric chemistry models must incorporate a greater range of HONO sources, especially in simulations of polluted atmospheres.

#### 4. Summary and conclusions

This research reported HONO and relevant chemical constituents and meteorological parameters at a coastal site in southern China during the autumn in 2018. The results provide insights into direct HONO emissions from ships and show significant contrasts in pollution characteristics,



**Fig. 8.** Comparisons of averaged model-simulated diurnal  $\cdot OH$  concentration (upper panel) and atmospheric oxidising capacity (AOC) by  $\cdot OH$  (lower panel) between two model runs with/without (red/black) constraints of measured HONO in marine (left), coastal (middle) and continental (right) air. The differences present impacts of HONO sources except for the  $NO + \cdot OH$  reaction on the modelling results. Shaded areas mark standard deviations ( $\pm 1\sigma$ ).

HONO budgets and radical chemistry in air masses originating from the sea, coastline and continent. An average ship HONO/NO<sub>x</sub> emission ratio of  $1.21 \pm 0.99\%$  was obtained from hundreds of night-time fresh ship plumes. HONO budget calculations suggest that the gas-phase NO + •OH reaction is the dominant HONO source in ship emission-perturbed marine air and that the sea surface likely serves as an HONO sink. Heterogeneous NO<sub>2</sub> conversion on underlying surfaces and surface-adsorbed HNO<sub>3</sub> photolysis were the main sources of HONO in coastal air, and heterogeneous NO<sub>2</sub> conversion on the ground surface as well as homogeneous NO + •OH reaction dominated HONO formation in continental air. HONO photolysis contributed significantly to RO<sub>x</sub> radicals in all three types of air, and its contribution to RO<sub>x</sub> and AOC increased from marine to coastal to continental air. This study provides new information on HONO sources and impacts in the polluted coastal boundary layers and highlights the necessity of comprehensively considering HONO sources in air quality and climate change modelling.

### CRedit authorship contribution statement

**Rongrong Gu** carried out the data analysis and wrote the manuscript. **Weihao Wang, Xiang Peng, Men Xia** and **Ya'nan Wang** conducted the field observations and data curation. **Min Zhao** and **Yingnan Zhang** performed the box model simulations. **Yiming Liu** plotted some figure. **Hengqing Shen** participated in the research discussion. **Tao Wang** and **Likun Xue** designed the study and revised the manuscript. **Wenxing Wang** provided resources.

### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.scitotenv.2022.153692>.

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